

Radiation Balance for Earth

Recap:

- Solar radiation at TOA is ~ blackbody at 5780 K
- Has small solid angle \rightarrow plane wave
- Integrated energy on a normal to the surface is $\sim 1370 \text{ W/m}^2$
- varies with latitude and season

We can calculate effective T of Earth by considering radiative equilibrium between solar and terrestrial radiation.

Energy absorbed by Earth

$$= \text{energy intercepted} \times (1 - \text{average reflection coefficient}) \\ = \pi R^2 F (1 - A)$$

where

R = Earth's radius

F = solar flux density at Earth

A = planetary albedo (= ratio of reflected flux density to incoming solar flux density)

Energy radiated by Earth

= emissivity \times surface area \times blackbody flux density

$$= \epsilon 4\pi R^2 \sigma T_e^4$$

$\epsilon \approx 1$ for infrared

$$\therefore 4\pi R^2 \sigma T_e^4 = \pi R^2 F(1-A)$$

$$T_e = \left[\frac{F(1-A)}{4\sigma} \right]^{1/4}$$

defines the effective radiating temperature of Earth (or any planet)

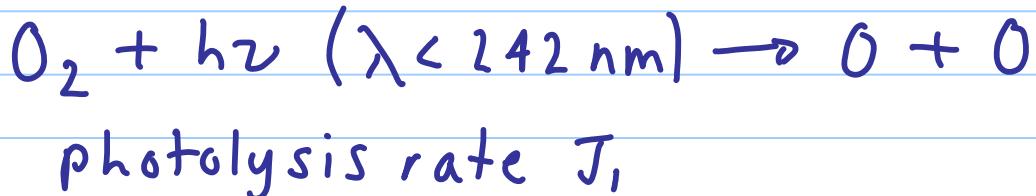
For Earth, $T_e = 255$ K

The atmosphere has a warming effect that makes the Earth's surface temperature about 33 K warmer than T_e .

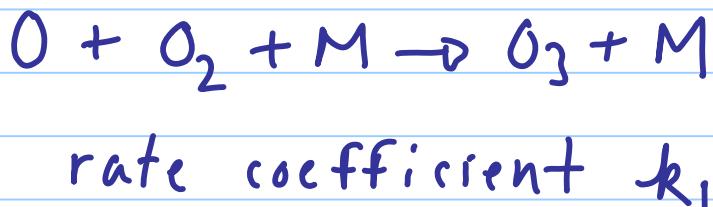


Oxygen and Ozone in the Atmosphere

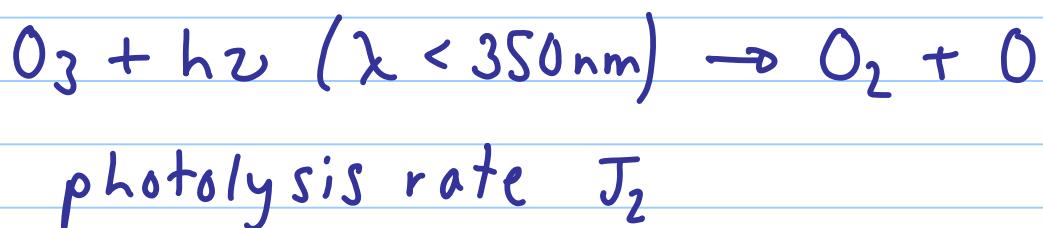
- ① atomic oxygen created by photolysis of molecular oxygen



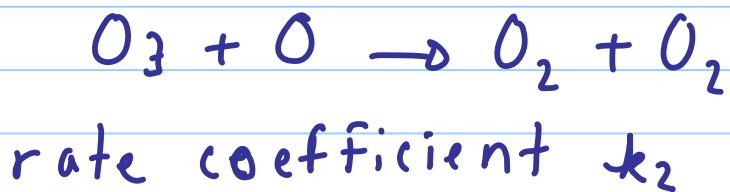
- ② production of ozone



- ③ photolysis of ozone



- ④ collision of ozone with atomic oxygen



- ⑤ self reaction of atomic oxygen
- $$O + O + M \rightarrow O_2 + M$$
- } very slow, can ignore
at < 60 Km

These reactions occur simultaneously
 \rightarrow Chapman Cycle

Reactions ② and ③ are faster than ① and ④, so for equilibrium of O and O_3 (odd oxygen) :

$$k_1 [O][O_2][M] = J_2 [O_3]$$

$$k_2 [O][O_3] = J_1 [O_2] \text{ for overall balance.}$$

Combine these :

$$[O] = \frac{J_1 [O_2]}{k_2 [O_3]}$$

$$[O_3] = \frac{k_1}{J_2} [O][O_2][M]$$

$$= \frac{k_1}{J_2} \frac{J_1 [O_2]}{k_2 [O_3]} [O][M]$$

$$\therefore [O_3] = [O_2] \sqrt{\frac{J_1 k_1 [M]}{J_2 k_2}}$$

So $[O_3]$ depends on J_1/J_2 , which increases with height.

$$J = \int_{\lambda} \sigma_{\lambda} F_{\lambda}(00) \tau_{\lambda}(z, 00) d\lambda$$

→ dissociating quanta per molecule that are absorbed

$[O_3]$ also depends on $[M]$ and $[O_2]$ which decrease with height.

∴ There is maximum in $[O_3]$ at some height
→ ozone layer

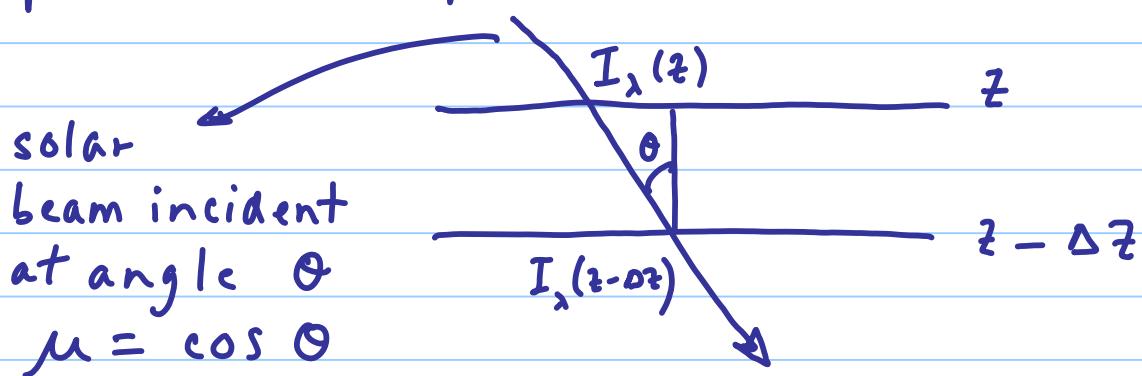
There will also be a maximum in heating of the atmosphere at this height

→ stratosphere



Solar Heating of the Atmosphere

Consider an atmospheric layer in a plane-parallel atmosphere.



$$I_\lambda(z - \Delta z) = I_\lambda(z) \tau_\lambda(\Delta z/\mu)$$

$$= I_\lambda(z) \exp\left(-k_s f \frac{\Delta z}{\mu}\right)$$

using

Schwarzschild's Equation for a homogeneous layer with $J=0$.

Energy lost (per unit area)

$$= [I_\lambda(z) - I_\lambda(z - \Delta z)] \Delta \Omega_s$$

$$= [I_\lambda(z) - I_\lambda(z) \tau_\lambda(\Delta z/\mu)] \Delta \Omega_s$$

$$= I_\lambda(z) A(\Delta z/\mu) \Delta \Omega_s \quad \text{where } A_\lambda = 1 - \tau_\lambda$$

$$= \frac{\Delta E_\lambda}{\Delta t}$$

This energy loss from the beam is also the net energy gain of the atmosphere, so

$$\frac{\Delta E_\lambda}{\Delta t} = \rho C_p \Delta V \frac{dT}{dt}$$

where ρ = density

$\rightarrow \text{J/Kg/K}$

C_p = specific heat at constant pressure

ΔV = volume of cylinder where heat is absorbed

$\Delta V = \Delta z / \mu$ assuming a cylinder of unit cross-section aligned along the beam

$$\begin{aligned}\frac{dT}{dt} &= \frac{1}{g C_p \Delta z / \mu} \frac{\Delta E}{\Delta t} \\ &= \frac{\mu}{g C_p \Delta z} I_\lambda(z) A_\lambda (\Delta z / \mu) \Delta R_s\end{aligned}$$

From Hydrostatic Eqn $\Delta p = -\rho g \Delta z$

$$\frac{dT}{dt} = -\frac{g}{C_p} \frac{\mu I_\lambda(p)}{\Delta p} A_\lambda (\Delta p / \mu) \Delta R_s$$

$$\frac{dT}{dt} = -\Gamma \frac{dF_\lambda(\mu, p)}{dp} = \text{heating rate}$$

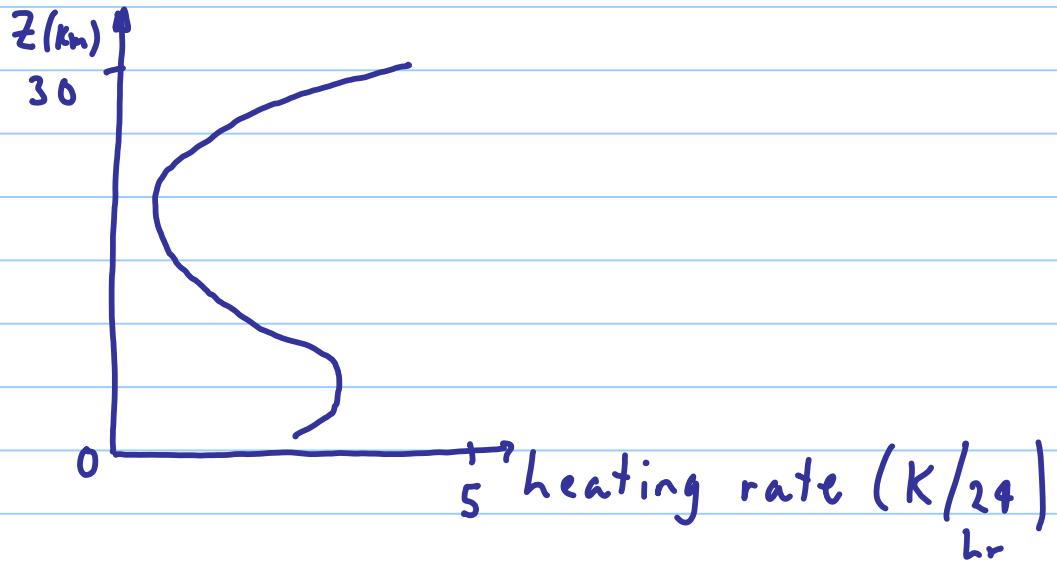
where

$$\Gamma = g / C_p = \text{adiabatic lapse rate}$$

$$F_\lambda = \mu I_\lambda(p) A_\lambda (\Delta p / \mu) \Delta R_s = \text{radian flux density incident on unit area}$$

Since dF_λ/dp is always < 0 , so $\frac{dT}{dt}$ is > 0 .

This equation allows calculation of the heating rates in the atmosphere due to solar radiation.



Plotting the heating rate for solar radiation in the lower atmosphere in $^{\circ}\text{C}/\text{day}$ shows:

- peak at the ground due to H_2O and CO_2 absorption
- then a decrease to a minimum
- above the tropopause, O_3 absorption becomes important, causing a large increase in heating rate

The above calculation of dT/dt is for monochromatic radiation.

Total heating rate due to solar radiation is

$$\left(\frac{dT}{dt} \right)_{\text{total}} = \sum_{\lambda} \left(\frac{dT}{dt} \right)_{\lambda}$$

$$= -\Gamma \sum_{\lambda_i} \frac{dF_{\lambda_i}}{dp} \Big|_p \Delta \lambda_i$$

sum
 over all
 spectral
 intervals
 "i"