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RESEARCH LETTER

Key Points:

- First long-term time series of NH₂ concentrations in the high Arctic
- Boreal wildfire emissions of NH₃ detected in Canadian Arctic and Southern Canada
- 2014 Northwest Territories fires were a considerable source of NH₃ in the summertime Arctic

Supporting Information:

Supporting Information S1

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Long-range transport of NH₃, CO, HCN, and C₂H₆ from the 2014 Canadian Wildfires

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Abstract We report the first long-term measurements of ammonia (NH_3) in the high Arctic. Enhancements of the total columns of NH_3 , carbon monoxide (CO), hydrogen cyanide (HCN), and ethane (C₂H₆) were detected in July and August 2014 at Eureka, Nunavut, and Toronto, Ontario. Enhancements were attributed to fires in the Northwest Territories using the FLEXPART Lagrangian dispersion model and the Moderate Resolution Imaging Spectroradiometer Fire Hot Spot data set. Emission estimates are reported as average emission factors for HCN (0.62 ± 0.34 g kg⁻¹), C₂H₆ (1.50 ± 0.75 g kg⁻¹), and NH₃ (1.40 ± 0.72 g kg⁻¹). Observations of NH₃ at both sites demonstrate long-range transport of NH₃, with an estimated NH₃ lifetime of 48 h. We also conclude that boreal fires may be an important source of NH₃ in the summertime Arctic.

1. Introduction

The Arctic has few sources of ammonia (NH₃), with the main sources being seabird guano [Blackall et al., 2007] and seal excreta [Theobald et al., 2006]. Although biomass burning has been identified as a significant source of NH₃ [Bouwman et al., 1997], its short lifetime, typically under 24 h, would make long-range transport from lower latitudes to the Arctic unlikely [Lefer et al., 1999]. Emissions of NH₃ from boreal fires in Canada and Alaska have been previously measured by aircraft-based in situ instruments [Nance et al., 1993] and airborne Fourier transform infrared (FTIR) spectroscopy [Goode et al., 2000]. Satellite observations have also quantified boreal fire emissions of NH₃, including Tropospheric Emission Spectrometer (TES) measurements of NH₃ from Alaskan fires [Alvarado et al., 2011] and Infrared Atmospheric Sounding Interferometer (IASI) measurements from the 2010 Russian fires [R'Honi et al., 2013]. These studies have attempted to quantify NH₃ emissions from boreal fires; however, possible long-range transport of NH₃ to the Arctic has not yet been investigated.

Atmospheric NH₃ reacts rapidly with acidic gases in the atmosphere leading to the formation of particulate matter [Behera et al., 2013; Hertel et al., 2012] which contribute to a reduction in air quality [Pope et al., 2009; Erisman et al., 2011]. NH₃ is also known to contribute to the acidification and eutrophication of soil and water bodies therefore affecting biodiversity [Bobbink et al., 1998, 2010], while in high concentrations NH₂ is toxic to living organisms. Transport of NH₃ to the Arctic will thus affect air quality, while deposition of NH₃ could influence biodiversity in this sensitive ecosystem. However, the effects of NH₃ in the Arctic remain unknown as its abundance, sources, and sinks have not yet been quantified in this region.

Measurement of NH₃ is difficult due to its low ambient concentrations and episodic emissions. Although satellite observations, such as those from TES and IASI, provide global coverage, NH₃ measurements are limited in high-latitude regions due to insufficient thermal contrast and low concentrations, even during fire-enhanced periods. The first measurements of NH₃ in the Canadian Arctic were performed by Wentworth et al. [2016], who made shipborne measurements of boundary layer NH₃ from 13 July to 7 August 2014. During this period, boreal fires were burning in the Northwest Territories, although the main source of NH₃ was attributed to seabirds. It is likely that shipborne measurements have limited sensitivity to biomass burning emissions as these would be more abundant in the free troposphere than in the boundary layer. For this reason, total column measurements of NH₃ are preferable for quantifying fire emissions of NH₃.

It has been recently shown that solar absorption spectroscopy using FTIR spectrometers can be used to measure NH₃ total columns [Dammers et al., 2015], while biomass burning emissions of carbon monoxide (CO), hydrogen cyanide (HCN) and ethane (C_2H_6) have been previously measured in the Canadian Arctic at Eureka [Viatte et al., 2013, 2014, 2015], Australia [Paton-Walsh et al., 2005, 2010], Reunion Island [Vigouroux et al., 2012],

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and Japan [*Zhao et al.*, 2002]. The ground-based FTIR instrument at Eureka provides measurements of NH₃ where satellite observations and in situ data are sparse or nonexistent.

In this paper, two FTIR instruments at Eureka, Nunavut, and Toronto, Ontario, were used to detect NH₃ emissions from the 2014 Northwest Territories fires. The fire source near Great Slave Lake (61.67°N, 114.00°W) is approximately 2300 km and 3100 km from Eureka and Toronto, respectively. NH₃ was measured at both sites, with detection of total column enhancements 2–5 times the background during the burning period. Simultaneous enhancements of CO, HCN, and C₂H₆ on the order of 1.5–2 times the background were also observed, providing further indication of long-range transport of fire emissions. The spatial separation between sites and differences in travel times of the smoke plume also allowed for an NH₃ lifetime of 48 h to be estimated. The enhancements at Eureka confirm the possibility of long-range transport of NH₃ and suggest that boreal fires may be a significant episodic source of NH₃ in the summertime Arctic.

2. Methods

2.1. Sites and Instruments

Trace gas column amounts are retrieved from solar absorption spectra obtained by two FTIR spectrometers: a Bruker IFS 125HR at the Polar Environment Atmospheric Research Laboratory (PEARL) in Eureka, Nunavut (80.05°N, 86.42°W) and an ABB Bomem DA8 at the University of Toronto Atmospheric Observatory (TAO) in Toronto, Ontario (43.66°N, 79.40°W). Measurements from both instruments are contributed to the Network for Detection of Atmospheric Composition Change (NDACC; http://www.ndsc.ncep.noaa.gov/). A detailed instrument description of the Bruker 125HR at PEARL is given by *Batchelor et al.* [2009]. Measurements are made in the midinfrared at a resolution of 0.0035 cm⁻¹ using a KBr beam splitter and either an InSb or HgCdTe detector and a series of optical filters. A detailed description of the Bomem DA8 at TAO is given by *Wiacek et al.* [2007] and *Whaley et al.* [2015]. Measurements are made with the same optical components and resolution as the PEARL instrument. For both instruments, spectra are recorded continuously during clear-sky daylight hours by alternating between optical filters while a heliostat system tracks the Sun.

2.2. Retrieval Methods

Solar absorption spectra were processed using the SFIT4 retrieval algorithm (https://wiki.ucar.edu/display/ sfit4/). SFIT4 is based on the Optimal Estimation Method [*Rodgers*, 2000] which iteratively adjusts trace gas volume mixing ratio profiles to minimize the difference between the measured and calculated spectra [*Pougatchev et al.*, 1995; *Rinsland et al.*, 1998]. The forward model of SFIT4 is a line-by-line radiative transfer model that uses a Voigt line shape. Temperature and pressure profiles are provided by the National Centers for Environmental Prediction (NCEP) and spectroscopic parameters from HITRAN 2008 [*Rothman et al.*, 2009] are used.

CO, HCN, and C_2H_6 are standard NDACC Infrared Working Group (IRWG, https://www2.acom.ucar.edu/irwg) products. These species were retrieved using IRWG-recommended microwindows and a priori profiles. For each site, 40 year average (1980–2020) profiles from the Whole-Atmosphere Community Climate Model (WACCM V6) [*Eyring et al.*, 2007] were used as the a priori profiles for these species. Three microwindows were used for CO: a strong line at 2157.51–2159.14 cm⁻¹ and two weak lines at 2057.68–2058.00 and 2069.56–2069.76 cm⁻¹ [*Notholt et al.*, 2000; *Zhao et al.*, 2002]. For HCN, three microwindows were used: 3268.00–3268.38, 3287.00–3287.48, and 3299.40–3299.60 cm⁻¹ [*Notholt et al.*, 2000; *Zhao et al.*, 2002; *Mahieu et al.*, 1997]. Three microwindows were also used for C₂H₆: 2976.66–2976.95 [*Mahieu et al.*, 1997; *Rinsland et al.*, 2002; *Paton-Walsh et al.*, 2010], 2983.20–2983.55 [*Meier et al.*, 2004], and 2986.50–2986.95 cm⁻¹ [*Notholt et al.*, 1997].

The microwindows used for NH₃ correspond to the absorption lines of the v_2 vibrational band [Dammers et al., 2015]. For Eureka, three microwindows were used: 929.40–931.40, 950.02–952.20, and 962.10–970.00 cm⁻¹. The first and last microwindows for Eureka were also used for Toronto but narrowed to 930.32–931.32 and 966.97–967.68 cm⁻¹, respectively, due to water vapor interference. The 950.02–952.20 cm⁻¹ microwindow was not used for Toronto as a result of saturation due to water vapor. The use of three microwindows at Eureka was necessary in order to achieve adequate information content due to the low NH₃ concentrations. For both sites, the NH₃ a priori profile was derived from balloon-based measurements [Toon et al., 1999] taken near Fairbanks, Alaska (64.84°N, 147.72°W).

For all retrievals, full error analysis was performed following *Rodgers* [2000]. The retrieval uncertainties include smoothing error due to the limited vertical resolution, forward model parameter error, and measurement

noise error. Adding these in quadrature, average uncertainties for the retrieved total columns of CO, HCN, C_2H_6 , and NH_3 are 3.4%, 10.3%, 1.1%, and 26.7%, respectively, for Eureka and 3.2%, 6.4%, 5.4%, and 20.2%, respectively, for Toronto. The average degrees of freedom for signal for CO, HCN, C_2H_6 , and NH_3 are 2.4, 1.8, 1.3, and 1.1, respectively, for Eureka and 3.5, 2.2, 1.9, and 1.1, respectively, for Toronto. For CO, HCN, and C_2H_6 , the total column averaging kernels (shown in the supporting information) show peak values in the upper troposphere at both sites, while the total column averaging kernels of NH_3 peak near the surface and decrease upward to 5 km. Differences in the total column averaging kernels between species were not taken into account in the analysis below, as CO, HCN, and C_2H_6 have similar vertical sensitivities. However, for NH_3 , the contribution to its total column from altitudes above 5 km could be underestimated as a result of the lack of vertical sensitivity in this region.

3. Results and Discussion

3.1. FTIR Time Series

The total column time series for CO, HCN, C_2H_6 , and NH_3 at Eureka and Toronto are shown in Figure 1. All years (2006–2014 for Eureka and 2002–2014 for Toronto) are shown in gray, while 2014 measurements are highlighted in red for Eureka and blue for Toronto. The maximum monthly mean CO total columns are observed in March at Eureka (2.07×10^{18} molecules cm⁻²) and April at Toronto (2.41×10^{18} molecules cm⁻²). Minimum monthly mean total columns are observed in September at Eureka (1.59×10^{18} molecules cm⁻²) and Toronto (1.85×10^{18} molecules cm⁻²). The main sources of CO are fossil fuel combustion and oxidation of volatile organic compounds and CH₄, as well as biomass burning emissions [*Holloway et al.*, 2000]. The primary sink of CO is oxidation by hydroxyl (OH) which contributes to the seasonal cycle at both sites. Larger total columns of CO are observed for Toronto due to nearby anthropogenic sources. Transport of CO from midlatitudes also contributes to the seasonal cycle at Eureka, while biomass burning emissions are observed in the summer months.

Maximum monthly mean total columns of HCN are observed in August at Eureka (6.8×10^{15} molecules cm⁻²) and July at Toronto (6.79×10^{15} molecules cm⁻²). Since HCN is relatively inactive, it serves as a good tracer of biomass burning [*Rinsland et al.*, 2001]. Additional sources are the result of emissions from plants, bacteria, and fungi due to increased vegetative activity in the springtime [*Cicerone and Zellner*, 1983]. The lifetime of HCN in the troposphere is approximately 5 months [*Li et al.*, 2003], with main sinks due to OH and O(¹D) reaction [*Cicerone and Zellner*, 1983] and ocean uptake [*Li et al.*, 2003]. Large variability of HCN is observed in the summer months at Eureka corresponding to the boreal fire season from May to August [*Macias Fauria and Johnson*, 2008]. Particularly, large fire events in Russia during July 2008 and August 2010 are observed in the time series [*Viatte et al.*, 2013, 2015] in addition to the 2014 Northwest Territories fires.

The sources of C_2H_6 are natural gas and fossil fuel emissions [*Singh and Zimmerman*, 1992] in addition to biomass burning, while its main sink is reaction with OH. The seasonal cycle of C_2H_6 is similar to that of CO due to their common sources and sinks. Maximum monthly mean total columns are observed in March at Eureka (2.67×10^{16} molecules cm⁻²) and February at Toronto (2.97×10^{16} molecules cm⁻²) with minima in August at Eureka (1.2×10^{16} molecules cm⁻²) and Toronto (1.49×10^{16} molecules cm⁻²). Both CO and C_2H_6 have relatively long atmospheric lifetimes at 52 days [*Daniel and Solomon*, 1998] and 80 days [*Xiao et al.*, 2008], respectively.

The time series of NH₃ at Eureka and Toronto show maximum monthly mean total columns in the summer months $(2.79 \times 10^{14} \text{ molecules cm}^{-2} \text{ in July}$ at Eureka and $9.80 \times 10^{15} \text{ molecules cm}^{-2}$ in May at Toronto) as a result of its biogenic sources, biomass burning, and agricultural emissions. The lower abundance of NH₃ at Eureka is due to its isolation from nearby sources. However, it has been suggested that seabird colonies may be a considerable source of NH₃ in the Arctic [*Wentworth et al.*, 2016]. For both sites, NH₃ is highly variable due to its short lifetime and its emissions being rather episodic in nature. For both Eureka and Toronto, large enhancements of NH₃ are observed in July and August 2014 due to boreal fires in the Northwest Territories as shown in section 3.2.

3.2. Source Attribution

Enhancements in the time series of each species were first identified by measurements with a total column greater than a 3σ deviation from the monthly mean taken over all years. By this criterion, three main enhancements were found for each site in July and August 2014. The first initial detection for Eureka occurred on 25 July 2014 (14:40 UTC) which corresponds to the maximum CO total column measured on that day. Two later enhancements were also detected at Eureka on 7 August 2014 (17:20 UTC) and 19 August 2014 (19:15 UTC).

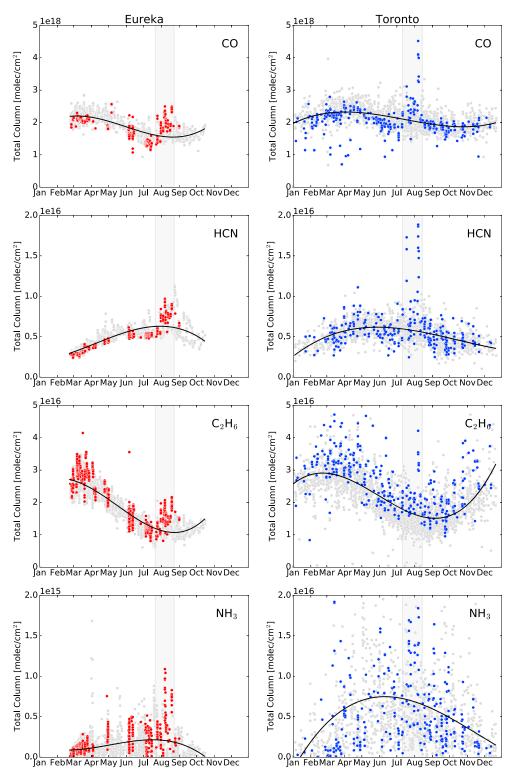


Figure 1. Total column time series of CO, HCN, C_2H_6 , and NH_3 for (left) Eureka (2006–2014) and (right) Toronto (2002–2014). All years are shown in gray, while 2014 is highlighted. The gray vertical bar indicates periods of fire-affected measurements (25 July to 22 August for Eureka and 11 July to 13 August for Toronto). The black line represents a polynomial fit to the data for all years.

Similarly, the initial enhancement at Toronto was found to have occurred on 11 July 2014 (14:09 UTC) with two later maxima on 18 July 2014 (19:05 UTC) and 7 August 2014 (17:20 UTC).

To determine the source of the observed enhancements, the FLEXPART [*Stohl et al.*, 2005] Lagrangian dispersion model was used. Backward dispersion runs were initialized by releasing an ensemble of 60,000 air-tracer particles over a 6 h period about the observed peak enhancement for CO. The particles were released over a $3^{\circ} \times 3^{\circ}$ box centered on each site. The model was run backward in time for 7 days driven by meteorological data from the NCEP Climate Forecast System (CFS V2) 6 h product [*Saha et al.*, 2011]. The FLEXPART model was run for each of the three peak enhancements observed at both sites.

The sensitivity of the measurement to various source regions is proportional to the residence time of the air-tracer particles. The FLEXPART sensitivities are shown in Figure 2 for both sites. The Moderate Resolution Imaging Spectroradiometer (MODIS) Active Fire data [*Giglio et al.*, 2006] provide the locations of active fires as shown in Figure 2. The FIRMS (Fire Resource Management System, https://earthdata.nasa.gov/earth-observation-data/near-real-time/firms/active-fire-data) fire product was used, where the plotted points correspond to fire detections with a confidence ratio of 0.75 or greater. For all FLEXPART backward runs, sensitivity to the Northwest Territories fires is observed. The initial enhancements on 25 July 2014 for Eureka and 11 July 2014 for Toronto show the least sensitivity to these fires, while the greatest sensitivity is observed in the 7 August 2014 FLEXPART runs for both sites, which is consistent with the largest enhancements observed at that time.

3.3. Trace Gas Correlations

The enhancement ratio (EnhR) is used to quantify emissions from biomass burning events for periods of fire-affected measurements. Fire-affected measurements were first classified as those with total column amounts greater than a 1 σ standard deviation from the monthly mean over all years. Since each species was retrieved in a different spectral region measured using optical filters, the enhancement ratio was limited to measurements occurring within 1 h of a CO measurement, where each CO measurement was only used once in the correlation. The correlations of HCN, C_2H_6 , and NH₃ with CO for Eureka and Toronto are shown in Figure 3. The colored points represent fire-affected measurements from 25 July to 22 August 2014 for Eureka (red) and 11 July to 13 August 2014 for Toronto (blue). For C_2H_6 , there is a linear trend for all measurements as a result of the similar seasonal cycles of CO and C_2H_6 . No trend is observed for HCN and NH₃ with CO due to their differing seasonal cycles.

The unified least squares procedure of *York et al.* [2004], which accounts for errors in both the ordinal and abscissa coordinates, was used to determine a linear regression for the fire-affected measurements. The result of the linear regression is shown as the dashed line in Figure 3, the slope of which is the enhancement ratio of the target species. The uncertainty of the enhancement ratio is the standard error of the slope given by *York et al.* [2004]. There is generally a good correlation with CO for HCN and C_2H_6 with correlation coefficients ranging from 0.70 to 0.93. For NH₃, the correlation coefficient is 0.70 for Eureka and 0.45 for Toronto. The lower correlation for Toronto NH₃ measurements is likely due to the variability of the total column amounts of NH₃ due to nearby agricultural and anthropogenic sources.

Since measurements are made at a distance from the fire source, the measured smoke plumes have experienced aging, resulting in some loss of each species. Aging of the plume is accounted for by calculation of the emission ratio (ER) at the fire source assuming a first-order loss of each species. The emission ratio is defined by

$$\mathsf{ER}_{X} = \mathsf{EnhR}_{X} \cdot \frac{\exp\left(\frac{t}{\tau_{X}}\right)}{\exp\left(\frac{t}{\tau_{CO}}\right)},\tag{1}$$

where τ_{χ} is the lifetime of the species and *t* is the travel time. Lifetimes for CO, HCN, and C₂H₆ were chosen to be 30, 75, and 45 days, respectively, following *Viatte et al.* [2015], which were determined by model comparisons to FTIR measurements. The travel times to each site were determined by HYbrid Single-Particle Lagrangian Integrated Trajectory (HYSPLIT) [*Rolph*, 2016; *Stein et al.*, 2015] back trajectories. For each of the three peak CO enhancements at each site described in section 3.2, the HYSPLIT model was run backward in time for 10 days. For each HYSPLIT model run, three trajectory altitudes between 3 and 10 km were used. The start time of each model run was adjusted to within 2 h of the observed peak CO enhancement, while the trajectory altitudes were also adjusted in order to provide back trajectories that were consistent with the fire source

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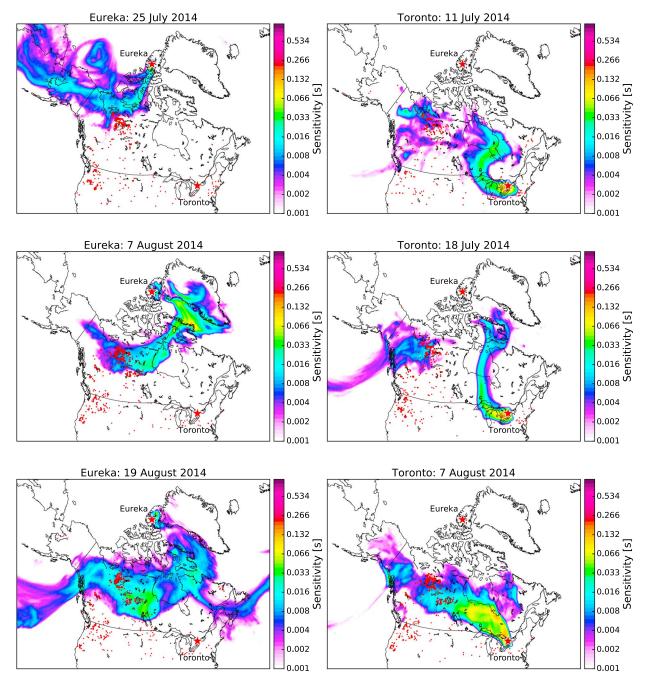


Figure 2. FLEXPART backward dispersion runs showing the total column sensitivity. MODIS fire hot spots are shown in red for 7 days prior to the release time. Particle release times for each panel correspond to the main CO enhancements observed at each site. Each panel represents a single FLEXPART model run, run backward in time for 7 days starting at the initial particle release.

regions identified by the FLEXPART model runs and MODIS fire data. From the three HYSPLIT model runs for each site, an average travel time was found to be 6 days for Eureka and 3 days for Toronto, with a standard deviation of 2 days and 1 day for Eureka and Toronto, respectively. The average travel times and standard deviation were determined from the three HYSPLIT runs over three trajectory altitudes. The emission ratios were calculated using these travel times and are shown in Table 1.

Because the travel time is dependent on the meteorological conditions along the trajectory of the smoke plume, it is difficult to account for variations in the travel time during the period of fire-affected measurements. For the long-lived species CO, HCN, and C_2H_6 , the variability of the travel times is small compared to

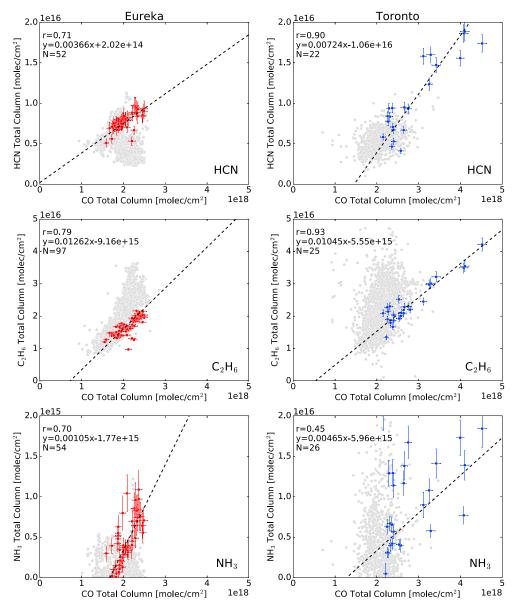


Figure 3. Enhancement ratios for HCN, C_2H_6 , and NH_3 for fire-affected measurements (25 July to 22 August 2014 for Eureka in red and 11 July to 13 August for Toronto in blue). The gray points represent all other measurements. The correlation coefficient *r*, linear equation of the fit, and number of fire-affected measurements are shown. Error bars correspond to the retrieval uncertainties for the respective species.

the lifetime of each species. It was found that the variability in the emission ratios for HCN and C_2H_6 due to the estimated variability of the travel times was within the uncertainties due to the enhancement ratios. For NH₃, the assumed lifetimes vary from 12 to 48 h and are comparable to the estimated travel times. Assuming this range of NH₃ lifetimes, the emission ratios were found to vary over several orders of magnitude for a single travel time. The variability of the emission ratio due to the estimated variability of the travel times for a single NH₃ lifetime was within this range. It was therefore concluded that the lifetime of NH₃ was the dominant source of uncertainty on the emission ratio. For these reasons, we have chosen the enhancement ratio uncertainty as the predominant uncertainty in the calculation of the emission ratio for all species.

For HCN, the emission ratios differ between Eureka (0.0037 \pm 0.0005) and Toronto (0.0072 \pm 0.0003) with a relative difference of approximately 82%, while a relative difference of 7% is found for the emission ratios of C₂H₆ at Eureka (0.0109 \pm 0.0004) and Toronto (0.0101 \pm 0.0005). The large difference in the HCN emission ratios suggests that additional sources of HCN were present for Toronto. This is evident in the FLEXPART sensitivities

•					. 2 0.	5		
			Eureka			Toronto		
Source	Platform	EF_{CO} (g kg ⁻¹)	EnhR	ER	EF (g kg ⁻¹)	EnhR	ER	EF (g kg ⁻¹)
			Н	ICN				
This study	Ground-based	127 (45) ^b	0.0037 (0.0005)	0.0032 (0.0005)	0.40 (0.15)	0.0072 (0.0003)	0.0068 (0.0003)	0.84 (0.30)
Akagi et al. [2011]	Compilation	127 (45)	_	_	1.52 (0.81)	_	_	1.52 (0.81)
Andreae and Merlet [2001] ^c	Compilation	107 (37)	-	_	0.81	_	_	0.81
<i>Viatte et al.</i> [2015] ^d	Ground-based	127 (45)	-	0.0034 (0.0009)	0.36 (0.17)	_	0.0034 (0.0009)	0.36 (0.17)
Goode et al. [2000] ^e	Aircraft	88.8	-	0.0069	0.61	_	0.0069	61
Simpson et al. [2011] ^f	Aircraft	112 (72)	-	0.0082 (0.0002)	0.89 (0.29)	_	0.0082 (0.0002)	0.89 (0.29)
Rinsland et al. [2007] ^g	Satellite	86 (17)	-	0.0024 (0.0003)	0.20 (0.05)	_	0.0024 (0.0003)	0.20 (0.05)
Tereszchuk et al. [2013] ^g	Satellite	-	-	0.0027 (0.0018)	-	-	0.0027 (0.0018)	-
			C	₂ H ₆				
This study	Ground-based	127 (45) ^b	0.0126 (0.0005)	0.0118 (0.0005)	1.61 (0.57)	0.0104 (0.0005)	0.0101 (0.0005)	1.38 (0.49)
Akagi et al. [2011]	Compilation	127 (45)	_	_	1.42 (0.43)	_	_	1.42 (0.43)
Andreae and Merlet [2001] ^c	Compilation	107 (37)	-	-	0.60 (0.15)	-	-	0.60 (0.15)
<i>Viatte et al</i> . [2015] ^d	Ground-based	127 (45)	_	0.0096 (0.0031)	1.09 (0.74)	_	0.0096 (0.0031)	1.09 (0.74)
Nance et al. [1993] ^e	Aircraft	81 (12)	-	0.0073 (0.0031)	0.66 (0.35)	_	0.0073 (0.0031)	0.66 (0.35)
Goode et al. [2000] ^e	Aircraft	88.8	_	0.0073	0.66	_	0.0073	0.66
Simpson et al. [2011] ^f	Aircraft	112 (72)	-	0.0046 (0.0009)	0.56 (0.13)	_	0.0046 (0.0009)	0.56 (0.13)
Rinsland et al. [2007] ^g	Satellite	86 (17)	_	0.0098 (0.0008)	0.91 (0.19)	_	0.0098 (0.0008)	0.91 (0.19)
Tereszchuk et al. [2013] ^g	Satellite	-	_	0.0069 (0.0023)	-	-	0.0069 (0.0023)	-
			٨	IH ₃				
This study ($\tau = 36$ h)	Ground-based	127 (45) ^b	0.0011 (0.0001)	0.0471 (0.0039)	3.64 (1.32)	0.0047 (0.0004)	0.0311 (0.0029)	2.40 (0.88)
This study ($\tau = 48$ h)	Ground-based	127 (45) ^b	0.0011 (0.0001)	0.0173 (0.0014)	1.34 (0.49)	0.0047 (0.0004)	0.0189 (0.0018)	1.46 (0.53)
Akagi et al. [2011]	Compilation	127 (45)	_	_	2.72 (2.32)	-	_	2.72 (2.32)
Andreae and Merlet [2001] ^c	Compilation	107 (37)	_	_	1.40 (0.80)	-	_	1.40 (0.80)
Nance et al. [1993] ^e	Aircraft	81 (12)	-	0.0130 (0.0050)	0.64 (0.31)	-	0.0130 (0.0050)	0.64 (0.31)
Goode et al. [2000] ^e	Aircraft	88.8	_	0.0158	0.86	-	0.0158	0.86
Alvarado et al. [2011] ^h	Satellite	-	-	0.0100 (0.0050)	-	-	0.0100 (0.0050)	-
R'Honi et al. [2013] ⁱ	Satellite	_	-	0.0310	-	-	0.0310	-

Table 1. Comparison of Calculated Enhancement Ratios, Emission Ratios, and Emission Factors for HCN, C₂H₆, and NH₃ at Eureka and Toronto^a

^aThe values in parentheses denote the reported uncertainties.

^bEF_{CO} for boreal forests from *Akagi et al.* [2011] used for conversion of ER to EF.

^cValues reported for extratropical forests.

^dGround-based FTIR measurements at Eureka, Nunavut.

^eMeasurements from Alaskan fires.

^fMeasurements from Canadian fires.

^gACE-FTS observations from Canadian and Alaskan fires.

^hTES observations from Canadian and Alaskan fires.

ⁱAverage value from IASI observations of the 2010 Russian fires.

of Figure 2. For each FLEXPART run for Toronto, considerable sensitivity to Northern Ontario and the Great Lakes region of Canada and the United States was observed. In addition, the emission ratios of HCN are an order of magnitude lower than those of C_2H_6 and the influence of additional sources is likely amplified.

Since NH₃ is short lived, plume aging has a significant effect on the measured enhancement ratio. To determine the emission ratios, four lifetimes for NH₃ were chosen: 12, 24, 36, and 48 h. The 12, 24, and 36 h lifetimes were chosen following *Whitburn et al.* [2015] from *Dentener and Crutzen* [1994], *Aneja et al.* [2001], and *R'Honi et al.* [2013], respectively. The 12 h and 24 h NH₃ lifetimes yielded large emission ratios (~10–100) and differed by an order of magnitude between sites. For this reason, these lifetimes were omitted from further analysis and it was inferred that the NH₃ lifetime must be considerably longer. Agreement was found assuming a 36 h lifetime with emission ratios of 0.0471 \pm 0.0039 and 0.0311 \pm 0.0029 for Eureka and Toronto, respectively.

The best agreement was found assuming a 48 h lifetime, which yielded emission ratios of 0.0173 ± 0.0014 and 0.0189 ± 0.0018 for Eureka and Toronto, respectively.

The assumption of a 48 h NH₃ lifetime is supported by comparison of the emission ratios to previously published values. Emission ratios for NH₃ of 0.0130 ± 0.0050 and 0.0158 were reported by *Nance et al.* [1993] and *Goode et al.* [2000], obtained from aircraft-based measurements of emissions from Canadian and Alaskan fires. The NH₃ emission ratio determined from TES observations of Canadian and Alaskan fires was 0.0100 ± 0.0050 [*Alvarado et al.*, 2011]. Using observations of the 2010 Russian fires from IASI, *R'Honi et al.* [2013] reported average NH₃ emission ratios ranging from 0.010 to 0.052. Our emission ratios are slightly larger than the values from aircraft and TES observations but agree with the lower limit from *R'Honi et al.* [2013]. Since the 2010 Russian fires included peatland burning [*Konovalov et al.*, 2011] in addition to boreal forest, the larger NH₃ emission ratios from *R'Honi et al.* [2013] are likely due to peatland burning.

The good agreement of the emission ratios of NH₃ between both sites and previous literature values assuming a 48 h lifetime suggests that the NH₃ lifetime is enhanced within a smoke plume. Bidirectional exchange of NH₃ between the atmosphere and the surface has been noted to increase NH₃ atmospheric concentrations and lifetimes [*Zhu et al.*, 2015]. However, the effects of bidirectional exchange are limited to the boundary layer and are of a different order of magnitude compared to the column enhancements due to transported biomass burning plumes. NH₃ may react rapidly with acidic gases to form ammonium particles [*Hertel et al.*, 2012], which could then be transported over large distances with lifetimes of 1-15 days [*Karlsson et al.*, 2013], therefore extending the spatial extent over which ammonia may be deposited. However, the abundance of other reactive trace gases and preexisting aerosol loads in a fire plume remains poorly known. It is possible that aerosol-gas exchange of NH₃ may extend the lifetime and long-range transport of NH₃ in a fire plume, although the magnitude of these effects is highly uncertain.

3.4. Emission Factors

Trace gas emissions from biomass burning are characterized by the emission factor. The emission factor is defined by [*Andreae and Merlet*, 2001]:

$$\mathsf{EF}_{\chi} = \mathsf{EF}_{\mathsf{CO}} \cdot \mathsf{ER}_{\chi} \cdot \left(\frac{\mathsf{MW}_{\chi}}{\mathsf{MW}_{\mathsf{CO}}}\right),\tag{2}$$

where MW is the molecular weight. To convert our emission ratios to equivalent emission factors, the emission factor of CO from *Akagi et al.* [2011] was used, which is based on a compilation of studies for boreal forest fuel types. Both *Akagi et al.* [2011] and *Andreae and Merlet* [2001] report emission factors for HCN, C_2H_6 , and NH_3 that can be compared to our calculated values (see Table 1). The values reported in *Andreae and Merlet* [2001] correspond to an extratropical fuel type that is a combination of boreal and temperate forests. Literature values for emission ratios and emission factors for boreal forests derived from ground-based, aircraft, and satellite platforms are also included in Table 1.

The emission factor for CO of 127 ± 45 g kg⁻¹ from *Akagi et al.* [2011] was used here to convert our calculated emission ratios to emission factors. As a result, differences in emission factors between sites are due to the same differences in the emission ratios discussed in section 3.3. For HCN, our emission factor for Eureka $(0.40\pm0.13 \text{ g kg}^{-1})$ is smaller than the compilation studies [*Akagi et al.*, 2011; *Andreae and Merlet*, 2001] and the aircraft-based studies [*Goode et al.*, 2000; *Simpson et al.*, 2011] but agrees within the combined uncertainties with the ground-based [*Viatte et al.*, 2015] and satellite [*Rinsland et al.*, 2007] studies. Similarly, our emission factors for C₂H₆ and those determined by ground-based [*Viatte et al.*, 2015] and satellite platforms [*Rinsland et al.*, 2007] are generally larger than that for aircraft studies [*Nance et al.*, 1993; *Goode et al.*, 2000]. For NH₃ with a lifetime of 48 h, our emission factors agree within the uncertainties with the compilation studies [*Andreae and Merlet*, 2001]. These differences are likely to be the result of lower emission factors of CO reported by *Nance et al.* [1993] and *Goode et al.* [2000] since the emission ratios of NH₃ were found to agree as stated in section 3.3.

4. Conclusions

The first long-term measurements of NH_3 in the Canadian Arctic have been presented here. Total columns of CO, HCN, C_2H_6 , and NH_3 were measured by ground-based FTIR spectrometers at Eureka, Nunavut, and Toronto, Ontario. Emission ratios for HCN, C_2H_6 , and NH_3 with respect to CO were determined for both sites.

The observed NH₃ enhancements at Eureka indicate that the 2014 Northwest Territories fires were a considerable episodic source of NH₃ to the Canadian Arctic. Simultaneous enhancements of CO, HCN, and C_2H_6 at Eureka, along with FLEXPART sensitivity runs, provided confirmation that the detected NH₃ enhancements originated from the Northwest Territories fires. Detection of simultaneous enhancements of all species at Toronto further demonstrated the long-range transport of NH₃ emissions from these fires. The consistency of the emission ratios for HCN, C_2H_6 , and NH₃ with respect to CO between the two sites and literature values, particularly for NH₃ with a estimated lifetime of 48 h, provides further confidence in these observations.

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